

### Lesson Plan

Name of College: GCW Lakhna Majra, Rohtak

Academic Session: 2024-25 Semester: Odd

Name of Asst./Ass.Prof: Anita Amani

Class: B.Sc. 1st sem (SEC)

Name of Subject: Chemistry

	<b>Analysis of soil: Composition of soil</b>
<b>22<sup>nd</sup>-27<sup>th</sup> July</b>	
<b>28<sup>th</sup> July</b>	<b>SUNDAY</b>
<b>29<sup>th</sup>-03<sup>rd</sup> August</b>	Concept of pH and pH measurement of soil
<b>04<sup>th</sup> August</b>	<b>SUNDAY</b>
<b>5<sup>th</sup>-10<sup>th</sup> August</b>	Complexometric titrations, Chelation, Chelating agents, use of indicators
<b>11<sup>th</sup> August</b>	<b>SUNDAY</b>
<b>12<sup>th</sup>-17<sup>th</sup> August</b>	Estimation of Calcium and Magnesium ions in soil.
<b>18<sup>th</sup> August</b>	<b>SUNDAY</b>
<b>19<sup>th</sup>-24<sup>th</sup> August</b>	Analysis of water: Definition of pure water, sources responsible for contaminating water, water sampling methods.
<b>25<sup>th</sup> August</b>	<b>SUNDAY</b>
<b>26<sup>th</sup>-31<sup>st</sup> August</b>	water purification methods. Determination dissolved oxygen of a water sample.
<b>1<sup>st</sup> September</b>	<b>SUNDAY</b>
<b>2<sup>nd</sup> -7<sup>th</sup> September</b>	A general study including preparation and uses of the Hair dye, soap, shampoo.
<b>08<sup>th</sup> September</b>	<b>SUNDAY</b>
<b>9<sup>th</sup>-14<sup>th</sup> September</b>	Preparation and uses of the suntan lotions, face powder, lipsticks, talcum powder, nail enamel.
<b>15<sup>th</sup> September</b>	<b>SUNDAY</b>
<b>16<sup>th</sup>-21<sup>st</sup> September</b>	General introduction to pesticides (natural and synthetic), benefits and adverse effects
<b>22<sup>nd</sup> September</b>	<b>SUNDAY</b>
<b>23<sup>th</sup>-28<sup>th</sup> September</b>	Changing concepts of pesticides, brief introduction of structure activity relationship
<b>29<sup>th</sup> September</b>	<b>SUNDAY</b>
<b>30<sup>th</sup> Sep -05<sup>th</sup> October</b>	Synthesis and technical manufacture and uses of representative pesticides in the Organochlorines (Gammexene,); Organophosphates (Malathion).
<b>06<sup>th</sup> October</b>	<b>SUNDAY</b>
<b>07<sup>th</sup>-12<sup>th</sup> October</b>	Basic principle of pH metric, potentiometric and conductometric titrations
<b>13<sup>th</sup> October</b>	<b>SUNDAY</b>
<b>14<sup>th</sup>-19<sup>th</sup> October</b>	Applications of conductivity measurements: determination of degree of dissociation
<b>20<sup>th</sup> October</b>	<b>SUNDAY</b>
<b>21<sup>st</sup>-26<sup>th</sup> October</b>	Determination of $K_a$ of acids and base, Buffer solution, Buffer action,
<b>27<sup>th</sup> October</b>	<b>SUNDAY</b>
<b>4<sup>th</sup> -9<sup>th</sup> November</b>	Henderson – Hazel equation, Buffer mechanism of buffer action.
<b>10<sup>th</sup> November</b>	<b>SUNDAY</b>
<b>11<sup>th</sup> -16<sup>th</sup> November</b>	Revision and test
<b>17<sup>th</sup> November</b>	<b>SUNDAY</b>
<b>18<sup>th</sup> November onwards till Exams.</b>	Test discussion

**LESSONPLAN-B.Sc5thSEMESTER****Session:2024-25**

Name of teacher-Anita Amani

Subject-Inorganic Chemistry

CLASS	WEEKS	SYLLABUS
B.Sc 5 <sup>th</sup> Semester	22-7-2024to27-7-2024	Metal-ligand Bonding in Transition Metal Complexes Limitations of valence bond theory,
	29-7-2024to3-8-2024	An elementary idea of crystal-field theory, crystal field splitting in octahedral
	5-8-2024to10-8-2024	Crystal field split tetrahedral and square planar complexes,
	12-8-2024to17-8-2024	Factors affecting the crystal-field parameters
	20-8-2024to24-8-2024	Thermodynamic and Kinetic Aspects of Metal Complex
	27-8-2024to31-8-2024	A brief outline of thermodynamic stability of metal complexes and factors affecting the stability,
	2-9-2024to7-9-2024	Substitution reactions of square planar complexes of Pt(II)
	9-9-2024to14-9-2024	Magnetic Properties of Transition Metal Complex Types of magnetic behaviour
	16-9-2024to21-9-2024	Methods of determining magnetic susceptibility ,spin-only formula.
	23-9-2024to28-9-2024	L-S coupling, correlation of s and f values, orbital contribution to magnetic moments ,application of magnetic moment data for 3dmetal complexes.
	30-9-2024to5-10-2024	Electron Spectra of Transition Metal Complexes Types of electronic transitions,
	7-10-2024to12-10-2024	Selection rules ford-d transitions ,spectroscopic ground states, spectrochemical series.
	14-10-2024to19-10-2024	Orgel-energy level diagram ford1 and d9 states,

	21-10-2024to26-10-2024	Discussion of the electronic spectrum of $[\text{Ti}(\text{H}_2\text{O})_6]^{3+}$ complex ion
	4-11-2024to9-11-2024	Revision of Electron Spectra, Doubts
	11-11-2024to20-11-2024	Assignments, Viva, Test, Revision
	23-11-2024to20-12-2024	MDU examination
	21-12-2024to31-12-2024	Winter break

**Lesson Plan for UG Classes B.Sc 3<sup>rd</sup> Year**

Assistant Professor: Dr. Naveen

Subject: Physical Chemistry

Paper Code: CH-502

Marks: 29+7

Month	WEEKS	SYLLABUS
July	4 <sup>th</sup> Week	Quantum Mechanic s-Black-body radiation, Plank's radiation law, photoelectric effect
	5 <sup>th</sup> Week	Heat capacity of solids, Compton effect, wave function and its significance of Postulates of quantum mechanics .,
August	1 <sup>st</sup> Week	Quantum mechanical operator, commutation relations, Hamiltonian operator, Hermitian operator, average value of square of Hermitian as a positive quantity,
	2 <sup>nd</sup> Week	Role of operators in quantum mechanics, To show quantum mechanically that position and momentum cannot be predicated simultaneously,
	3 <sup>rd</sup> Week	Determination of wave function & energy of a particle in one dimensional box, Pictorial representation and its significance
	4 <sup>th</sup> Week	Physical Properties and Molecular Structure Optical activity, polarization – (clausius – Mossotti equation). Orientation of dipoles in an electric field, dipole moment,
September	1 <sup>st</sup> Week	Measurement of dipole moment-temperature method and refractivity method, dipole moment and
	2 <sup>nd</sup> Week	Structure of molecules, Magnetic permeability, magnetic susceptibility and its determination. Application of magnetic susceptibility, magnetic properties – Para magnetism, diamagnetism and ferromagnetic.
	3 <sup>rd</sup> Week	Spectroscopy-I Introduction: Electromagnetic radiation, regions of spectrum, basic features of spectroscopy, statement of Born Oppenheimer approximation, Degrees of freedom..
	4 <sup>th</sup> Week	Rotational Spectrum Diatomic molecules. Energy levels of rigid rotator (semi-classical principles), selection rules,
October	1 <sup>st</sup> Week	Spectral intensity distribution using population distribution (Maxwell-Boltzmann distribution), determination of bond length, qualitative description of non-rigid rotor, isotope effect.
	2 <sup>nd</sup> Week	Vibrational spectrum Infrared spectrum: Energy levels of simple harmonic oscillator,
	3 <sup>rd</sup> Week	Selection rules, pure vibrational spectrum, intensity, determination of force constant and qualitative relation of force constant and bond energies,
	4 <sup>th</sup> Week	Effects of anharmonic motion and isotopic effect on the spectra., idea of vibrational frequencies of different functional groups.
November	1 <sup>st</sup> Week	Raman Spectrum: Concept of polarizability, pure rotational and pure vibrational Raman spectra of diatomic molecules, selectin rules, Quantum theory of Raman spectra
	2 <sup>nd</sup> Week	Assignments, Test, Revision
	3 <sup>rd</sup> Week	Test, Revision
	25/11/2024 - 20/12/2024	Theory examination
	21/12/2024- 31/12/2024	Winter break

**Lesson Plan for UG Classes B.Sc 3<sup>rd</sup> Year**

Assistant Professor: Dr. Naveen  
Paper Code: CH-503

Subject: Organic Chemistry  
Marks: 30+8

MONTH	WEEKS	SYLLABUS
July	4 <sup>th</sup> Week	NMR Spectroscopy-I Principle of nuclear magnetic resonance, the PMR spectrum, number of signals, peak areas,
	5 <sup>th</sup> Week	Equivalent and nonequivalent protons positions of signals and chemical shift, shielding and deshielding of protons,
August	1 <sup>st</sup> Week	Proton counting, splitting of signals and coupling constants, magnetic equivalence of protons
	2 <sup>nd</sup> Week	NMR Spectroscopy-II Discuss ion of PMR spectra of the molecules: ethyl bromide, n-propyl bromide, isopropyl bromide,
	3 <sup>rd</sup> Week	1,1-dibromoethane, 1,1,2-tribromoethane, ethanol, acetaldehyde, ethyl acetate, toluene,
	4 <sup>th</sup> Week	benzaldehyde and acetophenone. Simple problems on PMR spectroscopy for structure determination of organic compounds
September	1 <sup>st</sup> Week	Carbohydrates-I Classification and nomenclature. Monosaccharides, mechanism of osazone formation, interconversion of glucose and fructose, chain lengthening and chain shortening of aldoses.
	2 <sup>nd</sup> Week	Configuration of monosaccharides. Erythro and threo diastereomers. Conversion of glucose in to mannose. Formation of glycosides, ethers and esters
	3 <sup>rd</sup> Week	Determination of ring size of glucose and fructose. Open chain and cyclic structure of D (+)-glucose & D (-) fructose.
	4 <sup>th</sup> Week	Mechanism of mutarotation. Structures of ribose and deoxyribose
October	1 <sup>st</sup> Week	Carbohydrates-II An introduction to disaccharides (maltose, sucrose and lactose) and.
	2 <sup>nd</sup> Week	Polysaccharides (starch and cellulose) without involving structure determination
	3 <sup>rd</sup> Week	Organometallic Compounds, Organmagnesium compounds: the Grignard reagents-formation, structure and chemical reactions.
	4 <sup>th</sup> Week	Organozinc compounds: formation and chemical reactions.
November	1 <sup>st</sup> Week	Organolithium compounds: formation and chemical reactions.
	2 <sup>nd</sup> Week	Assignments, Test, Revision
	3 <sup>rd</sup> Week	Test, Revision
	25/11/2024 – 20/12/2024	Theory examination
	21/12/2024 – 31/12/2024	Winter break

**LESSON PLAN****TEACHER:** Dr. Heena**DEPARTMENT:** Department of Chemistry**SESSION:** 2024-2025**CLASS:** B.Sc. Life Sciences/Physical Sciences**SEMESTER:** I

<b>WEEK</b>	<b>TOPICS TO BE COVERED</b>
<b>WEEK 1</b> 22/07/2024 – 27/07/2024	Electronic displacements and its applications
<b>WEEK 2</b> 29/07/2024 – 03/08/2024	Reaction intermediates
<b>WEEK 3</b> 05/08/2024 – 10/08/2024	Aromaticity
<b>WEEK 4</b> 12/08/2024 – 17/08/2024	Isomerism
<b>WEEK 5</b> 19/08/2024 – 24/08/2024	Optical Isomerism, optical activity, symmetry elements, chirality
<b>WEEK 6</b> 26/08/2024 – 31/08/2024	Enantiomers, stereogenic centre, two stereocentres, diastereomers
<b>WEEK 7</b> 02/09/2024 – 07/09/2024	Threo-erythro diastereomers, meso compounds, resolution, inversion, retention, racemization
<b>WEEK 8</b> 09/09/2024 – 14/09/2024	Relative and absolute configurations, sequence rules, R and S nomenclature.
<b>WEEK 9</b> 16/09/2024 – 21/09/2024	Maxwell velocity and energy distribution, RMS, average, most probable velocity calculation.
<b>WEEK 10</b> 23/09/2024 – 28/09/2024	Collision diameter, number, frequency and mean free path. Real gas deviation, van der Waals equation of states
<b>WEEK 11</b> 30/09/2024 – 05/10/2024	Boyle's temperature, compression factor, behaviour of real gases.
<b>WEEK 12</b> 07/10/2024 – 12/10/2024	Critical temperature, pressure, volume determination. PV isotherms of real gases
<b>WEEK 13</b> 14/10/2024 – 19/10/2024	Continuity of states, van der Waal isotherms, relations of critical and van der Waal constants
<b>WEEK 14</b> 21/10/2024 – 26/10/2024	Compressibility factor, law of corresponding states.
<b>WEEK 15</b> 28/10/2024 – 02/11/2024	Diwali break
<b>WEEK 16</b> 04/11/2024 – 09/11/2024	Revision and tests
<b>WEEK 17</b> 11/11/2024 – 16/11/2024	Revision and tests
<b>WEEK 18</b> 18/11/2024 – 23/11/2024	Revision and tests
25/11/2024 – 20/12/2024	Theory examination
21/12/2024 – 31/12/2024	Winter break

**LESSON PLAN****TEACHER:** Dr. Heena**DEPARTMENT:** Department of Chemistry**SESSION:** 2024-2025    **CLASS:** B.Sc.    **SEMESTER:** III    **Paper:** CH-301,302

<b>WEEK</b>	<b>TOPICS TO BE COVERED</b>
<b>WEEK 1</b> 22/07/2024 – 27/07/2024	Thermodynamic terms, system types, extensive-intensive properties
<b>WEEK 2</b> 29/07/2024 – 03/08/2024	State and path functions, thermodynamic process, concept of heat and work, zeroth law of thermodynamics
<b>WEEK 3</b> 05/08/2024 – 10/08/2024	First law of thermodynamics, internal energy, enthalpy, heat capacity, heat capacities at constant volume and pressure
<b>WEEK 4</b> 12/08/2024 – 17/08/2024	Relation between $C_p$ and $C_v$ , Joule's law, Joule's coefficient, inversion temperature
<b>WEEK 5</b> 19/08/2024 – 24/08/2024	W, q, dU, dH calculation for ideal gas under isothermal, adiabatic conditions for reversible process. Enthalpy Temp. dependence
<b>WEEK 6</b> 26/08/2024 – 31/08/2024	Kirchoff's equation, bond energies and applications.
<b>WEEK 7</b> 02/09/2024 – 07/09/2024	Equilibrium constant, G, chemical potential, law of eq. derivation
<b>WEEK 8</b> 09/09/2024 – 14/09/2024	K dependence on temperature, reaction isochore, isotherm, Le-Chatelier's principle and applications
<b>WEEK 9</b> 16/09/2024 – 21/09/2024	Clausius Clapeyron equation and applications, Distribution law derivation, modifications of law
<b>WEEK 10</b> 23/09/2024 – 28/09/2024	Applications of distribution law, Werner's coordination theory, EAN concept.
<b>WEEK 11</b> 30/09/2024 – 05/10/2024	Chelates, Nomenclature, Isomerism, VBT of transition metal complexes
<b>WEEK 12</b> 07/10/2024 – 12/10/2024	Physical properties of a solvent
<b>WEEK 13</b> 14/10/2024 – 19/10/2024	Types of solvent and characteristics
<b>WEEK 14</b> 21/10/2024 – 26/10/2024	Reactions in liquid ammonia and sulphur dioxide
<b>WEEK 15</b> 28/10/2024 – 02/11/2024	Diwali break
<b>WEEK 16</b> 04/11/2024 – 09/11/2024	Revision and tests
<b>WEEK 17</b> 11/11/2024 – 16/11/2024	Revision and tests
<b>WEEK 18</b> 18/11/2024 – 23/11/2024	Revision and tests
25/11/2024 – 20/12/2024	Theory examination
21/12/2024 – 31/12/2024	Winter break

## Session 2024-25 Semester I (Odd)

### Lesson Plan for Arts/ Physical Sciences/ Life Sciences/ Commerce

Name of Program: UG Multi/ Interdisciplinary Program in Chemistry

Name of Course: Discipline Specific Course/ Major Course (DSC)

Nomenclature of Course: General Chemistry-I

Course Code: 24CHES401DS01

Credits (L+T+P): 2:0:0

Marks: 35+15=50

MONTH	WEEK	SYLLABUS
<b>July</b>	4 <sup>th</sup> week	Atomic Structure and Periodicity of Elements
	5 <sup>th</sup> week	Bohr's atomic model and its applications
<b>August</b>	1 <sup>st</sup> week	quantum numbers, their application and rules of electronic configuration,
	2 <sup>nd</sup> week	effective nuclear charge,
	3 <sup>rd</sup> week	shielding or screening effect, Slater rules,
	4 <sup>th</sup> week	variation of effective nuclear charge in periodic table.
	5 <sup>th</sup> week	Periodic trends in atomic radii,
<b>September</b>	1 <sup>st</sup> week	ionic radii and its calculation,
	2 <sup>nd</sup> week	covalent radii, electronegativity,
	3 <sup>rd</sup> week	electron gain enthalpy,
	4 <sup>th</sup> week	ionization enthalpy and factors affecting ionization energy
	5 <sup>th</sup> week	. Pauling, Mullikan and Allred Rachow scales
<b>October</b>	1 <sup>st</sup> week	Ionic Solids: Ionic bond and its characteristics and factors affecting, types of Bravais lattice, voids
	2 <sup>nd</sup> week	, packing in solids, determination of radius ratio of all voids, radius ratio rule and its limitations.
	3 <sup>rd</sup> week	Packing of ions in crystals, calculation of density and crystal structures of ionic solids (NaCl, CsCl, ZnS, CaF <sub>2</sub> , Na <sub>2</sub> O),
	4 <sup>th</sup> week	defect structures in crystal. Born-Landé equation with derivation, expression for lattice energy,
<b>November</b>	2 <sup>nd</sup> week	Madelung constant, Born-Haber cycle and its application with examples, solvation energy.
	3 <sup>rd</sup> week	Semiconductors, types of semiconductors, valence bond and band theories (alloys excluded).
	4 <sup>th</sup> week	Test and Revision

**Session 2024-25 Semester III (Odd)**

**Lesson Plan for Arts/ Physical Sciences/ Life Sciences/ Commerce**

Name of Program: UG Multi/ Interdisciplinary Program in Chemistry

Name of Course: Discipline Specific Course/ Major Course (DSC)

Nomenclature of Course: Organic Chemistry-3

MONTH	WEEK	SYLLABUS
<b>July</b>	4 <sup>th</sup> week	1. Alcohols Monohydric alcohols nomenclature, methods of formation by reduction of aldehydes, ketones, carboxylic acids and esters.
	5 <sup>th</sup> week	Hydrogen bonding. Acidic nature. Reactions of alcohols. Dihydric alcohols — nomenclature, methods of formation,
<b>August</b>	1 <sup>st</sup> week	chemical reactions of vicinal glycols, oxidative cleavage [Pb(OAc) <sub>4</sub> and HIO <sub>4</sub> ] and pinacol-pinacolone rearrangement.
	2 <sup>nd</sup> week	2. Epoxides Synthesis of epoxides.
	3 <sup>rd</sup> week	Acid and base-catalyzed ring opening of epoxides,
	4 <sup>th</sup> week	orientation of epoxide ring opening,
	5 <sup>th</sup> week	reactions of Grignard and organolithium reagents with epoxides Test and Revision
<b>September</b>	1 <sup>st</sup> week	Phenols Nomenclature, structure and bonding. Preparation of phenols, physical properties and acidic character.
	2 <sup>nd</sup> week	Comparative acidic strengths of alcohols and phenols, resonance stabilization of phenoxide ion.
	3 <sup>rd</sup> week	Reactions of phenols — electrophilic aromatic substitution, Mechanisms of Fries rearrangement,
	4 <sup>th</sup> week	Claisen rearrangement, Reimer-Tiemann reaction, Kolbe's reaction and Schotten and Baumann reactions
	5 <sup>th</sup> week	Test and Revision
<b>October</b>	1 <sup>st</sup> week	Ultraviolet (UV) absorption spectroscopy Absorption laws (Beer-Lambert law),
	2 <sup>nd</sup> week	molar absorptivity, presentation and analysis of UV spectra, types of electronic transitions, effect of conjugation
	3 <sup>rd</sup> week	. Concept of chromophore and auxochrome. Bathochromic, hypsochromic, hyperchromic and hypochromic shifts. UV spectra of conjugated dienes and enones,
	4 <sup>th</sup> week	Woodward-Fieser rules, calculation of max of simple conjugated dienes

		and , -unsaturated ketones. Applications of UV Spectroscopy in structure elucidation of simple organic compounds
<b>November</b>	2 <sup>nd</sup> week	Carboxylic Acids & Acid Derivatives Nomenclature of Carboxylic acids, structure and bonding, physical properties, acidity of carboxylic acids, effects of substituents on acid strength
	3 <sup>rd</sup> week	. Preparation of carboxylic acids. Reactions of carboxylic acids. Hell-Volhard-Zelinsky reaction. Reduction of carboxylic acids. Mechanism of decarboxylation. Structure , nomenclature and preparation of acid chlorides, esters, amides and acid anhydrides.
	4 <sup>th</sup> week	Relative stability of acyl derivatives. Physical properties, interconversion of acid derivatives by nucleophilic acyl substitution. Mechanisms of esterification and hydrolysis (acidic and basic) Test and Revision